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REACTIONS OF 1,2,3-TRIPHENYL-1,2,3-TRIPHOSPHAINDAN WITH TRIRUTHENIUM CLUSTER

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Reactions of 1,2,3-triphenyl-1,2,3-triphosphaindan (I), $C_6H_4(PPh)_3$, with $[Ru_3(CO)_{12}]$ under various conditions result in the rupture of the P_3C_2 -ring framework in the ligand and the cleavage of the M—M bonds in the parent cluster, leading to the formation of a series of phosphido bridged and phosphinidene capped trinuclear or polyhedral ruthenium clusters: a tetra-nuclear butterfly-like cluster $[Ru_4(CO)_{10}(\mu_3\text{-PPh})]$ 1, a tri-nuclear bent-chain skeleton cluster $[Ru_3(CO)_9\{\mu_3-\eta^3\text{-PPh}C_6H_4(PPh)_2\}]$ 2, and a nonplanar six-atom-raft rhombic geometry cluster $[Ru_6(CO)_{12}(\mu_3\text{-PPh})(\mu_4\text{-PPh})_2(\mu_3-\eta^2\text{-}C_6H_4)]$ 3. All compounds have been fully characterized by spectroscopic methods, while the molecular structures of the new compounds 2 and 3 are established by X-ray crystallographic techniques.

Keywords: Cluster; phosphorus; ruthenium; triphosphaindan

INTRODUCTION

The reactions of homocyclic phosphines $(PR)_n$ $(R = Ph,^{1a-c} Et,^{1c-e} CF_3,^{1f}$ or $Bu^t,^2$ n=4 or 5) with Group VIIIA metals (Fe, Ru, Os) carbonyl clusters have been extensively studied; however, less attention has been paid to the heterocyclic polyphosphines. For the investigated ligand 1,2,3-triphenyl-1,2,3-triphosphaindan (I), its reactions with Fe₂, Fe₃, and Os₃ carbonyl cluster have been reported by King³⁻⁴ and our group,⁵ respectively, to give a series of metal cluster derivatives with coordination of ligand (I) either intact or with the cleavage of one or two endocyclic phosphorus-phosphorus bonds, but without the observation of breakage of endocyclic P—C bonds. This paper reports an

in-depth investigation of the reactions of (\mathbf{I}) with triruthenium carbonyl clusters.

RESULTS AND DISCUSSION

The reaction of (I) with a two-fold molar amount of $[Ru_3(CO)_{12}]$ in toluene at $110^{\circ}C$ for 2 h gives rise to a tetra-nuclear butterfly cluster $[Ru_4(CO)_{10}(\mu_3\text{-PPh})]$ 1, and a tri-nuclear bent-chain cluster $[Ru_3(CO)_9\{\mu_3-\eta^3\text{-PPh}C_6H_4(PPh)_2\}]$ 2 in moderate yields (15% and 8%, respectively) after TLC separation using CH_2Cl_2 -hexane (1/4) as eluent. When the reaction temperature is raised to $150^{\circ}C$, a nonplanar six-atom-raft rhombic geometry cluster $[Ru_6(CO)_{12}(\mu_3\text{-PPh})(\mu_4\text{-PPh})_2(\mu_3-\eta^2\text{-}C_6H_4)]$ 3 is afforded in 7% yield. When an equimolar amount of (I) reacts with $[Ru_3(CO)_{12}]$ at $100^{\circ}C$, only cluster 2 is isolated and characterized. Cluster 1 has been earlier reported to be the product from the pyrolysis of $[Ru_3(CO)_9(\mu\text{-PPh}_2)(\mu\text{-H})_2]$.

The spectroscopic data of all the clusters obtained are given in Table I. The carbonyl stretching vibrations fall in the region between 2150 and 1900 cm⁻¹, indicating that all carbonyl groups are terminal. The ¹H NMR spectra only show the resonance signals for the phenyl protons.

TABLE I Spectroscopic Data for Clusters 1 to 3

Cluster	IR, $\nu(\mathrm{CO})^a/\mathrm{cm}^{-1}$	$^{1}\mathrm{H~NMR}^{b}$	$^{31}\mathrm{P}\{^{1}\mathrm{H}\}\ \mathrm{NMR}^{b}$
1	2097w, 2061vs, 2050m, 2042w, 2010w	6.5–7.8 (m, Ph)	88.7 (s)
2	2069s, 2046vs, 2015s, 2006s,	6.6–8.2 (m, Ph)	$112.5 (d, 272.0, P^1/P^2) \\ 27.6 (dd, 272.0, 20.4, P^1/P^2)$
3	1985w, 1975w 2080w, 2061s, 2038s, 2027s, 2002w	6.8–8.2 (m, Ph)	9.2 (d, 20.4, P ³) 87.2 (d, P ² and P ³) -51.2 (t, 272.0, 20.4, P ¹)

^aIn CH₂Cl₂.

^bIn CDCl₃ with SiMe₄ for ¹H and 85% H₃PO₄ for ³¹P as reference.

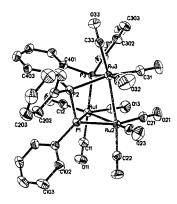


FIGURE 1 Molecular structure of **2** with H-atom omitted. Selected bond distances: Ru(1)-Ru(2) 2.8620(3); Ru(2)-Ru(3) 2.9456(3); Ru(1)...Ru(3) 3.903; Ru(1)-P(1) 2.3467(6); Ru(1)-P(3) 2.4041(6); Ru(2)-P(1) 2.3042(6); Ru(3)-P(2) 2.3552(6); Ru(3)-P(3) 2.3861(6); P(1)-P(2) 2.1806(8).

Molecular Structure of 2

The ${}^{31}P\{{}^{1}H\}$ NMR spectrum of **2** exhibits an AMX spin pattern, as indicated in Table I. From the ${}^{31}P$ NMR spectrum, we can deduce that the P_3C_2 -ring framework of the ligand (**I**) moiety in the cluster has been ruptured with the cleavage of one P—P bond. The elemental analysis results are in accordance with the formula $C_{33}H_{19}O_9P_3Ru_3$: C, 42.28(41.46); H, 2.35 (1.99); P, 9.02(9.74).

The molecular structure of **2** is shown in Figure 1, together with the atomic labeling scheme and the selected bond parameters listed below. Analogous structures containing Fe and Os have been reported before. ^{3b,4} The skeleton contains a bent-chain arrangement of triruthenium atoms with an angle of 84.450(7)°, which is supported by the ligand **I** fragments with one of the P–P bonds broken. The three Ru atoms and three P atoms define the vertices of a distorted trigonal prism with a nonbonded P–P edge linked to two carbon atoms of the fused phenyl ring.

Molecular Structure of 3

The molecular structure of **3** is illustrated in Figure 2, together with atomic labeling scheme and some selected bond parameters that are listed below. In the structure of **3**, six Ru atoms constitute a non-planar six-atom-raft rhombic (or two edge-fused butterflies) geometry skeleton, with the two internal triangles approximately coplanar (with dihedral angle of 165.8°) and each of the two external triangles nearly

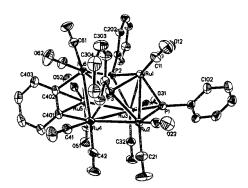


FIGURE 2 Molecular structure of **3** with H-atom omitted. Selected bond distances: Ru(1)-Ru(2) 2.8696(3); Ru(1)-Ru(3) 2.8650(3); Ru(2)-Ru(3) 2.8842(3); Ru(2)-Ru(4) 2.8136(3); Ru(3)-Ru(4) 3.0914(3); Ru(3)-Ru(5) 2.8083(3); Ru(4)-Ru(5) 2.7241(3); Ru(4)-Ru(6) 2.8774(3); Ru(5)-Ru(6) 2.7589(3).

perpendicular to its corresponding internal triangle (with dihedral angles of 78.3 and 79.4°, respectively). The six Ru atoms constitute a boat configuration. The two edge-fused butterfly geometry metal skeleton supports the ligands; a μ_3 -PPh, two μ_4 -PPh, and a μ_3 - η^2 -C₆H₄, which are formed by formal cleavage of the two P-P bonds and the two P-C(sp) bonds in the P_3C_2 -ring of the ligand (I). One of the external triangles (Ru(1)Ru(2)Ru(3)) is capped asymmetrically by the μ_3 -phosphinidene P(1) with P(1)-Ru(2) (2.2096(8) Å) a little shorter than the other two P-Ru bonds (average 2.2826(8) Å), while the other one is capped by the μ_3 - η^2 -benzyne, which derives from the formal cleavage of two P-C(sp) bonds in the fused phenyl ring of the ligand. The two μ_4 -phosphinidenes each caps the two wing-tip atoms Ru(1), Ru(6) and an edge of the base of the boat structure comprising the six Ru atoms. The Ru-Ru distances cover a wide range of 2.7241(3)-3.0914(3) Å. The longest bond is the diagonal of the Ru(2)Ru(3)Ru(4)Ru(5) rhomboid. The valence electron count for cluster 3 leads to 88, which is 2 electrons fewer than the EAN rule for the rhombic-raft geometry.

The $^{31}P\{^{1}H\}$ NMR spectrum exhibits two kinds of resonance signals with an integration intensity ratio of 2:1, which supports the X-ray structure. Elemental analytical results are in accordance with the formula $C_{36}H_{19}O_{12}P_{3}Ru_{6}$: C, 32.58(32.17); H, 1.67(1.41); P, 6.12(6.92).

REFERENCES

(a) H. G. Ang, L. L. Koh, and Q. Zhang, J. Chem. Soc., Dalton Trans., 2757 (1995);
 (b) H. G. Ang, S. G. Ang, and Q. Zhang, J. Chem. Soc., Dalton Trans., 3843 (1996);

- (c) H. G. Ang, S. G. Ang, and Q. Zhang, J. Chem. Soc., Dalton Trans., 2773 (1996); (d) H. G. Ang, S. G. Ang, W. L. Kwik, and Q. Zhang, J. Organomet. Chem., 485, C10 (1995); (e) H. G. Ang, S. G. Ang, and Q. Zhang, Phosphorus, Sulfur and Silicon Relat. Elem., 110, 145 (1996); (f) H. G. Ang, K. W. Aug, S. G. Ang, and A. L. Rheingold, J. Chem. Soc., Dalton Trans., 3131 (1996).
- (a) B. F. G. Johnson, T. M. Layer, J. Lewis, P. R. Raithby, and W. T. Wong, J. Chem. Soc.,
 Dalton Trans., 973 (1993); (b) E. Charalambous, L. Heuer, B. F. G. Johnson, J. Lewis,
 W. S. Li, M. McPartlin, and A. D. Massey, J. Organomet. Chem., 468, C9 (1994).
- [3] (a) R. B. King and R. H. Reimann, *Inorg, Chem.*, **15**, 184 (1976); (b) E. P. Kyba, K. L. Hassett, B. Sheikh, J. S. Mckennis, R. B. King, and R. E. Davis, *Organometallics*, **4**, 994 (1985).
- [4] S. G. Ang, X. H. Zhong, and H. G. Ang, J. Chem. Soc., Dalton Trans., 1151 (2001).
- [5] S. A. MacLaughlin, N. J. Taylor, and A. J. Carty, Can. J. Chem., 60, 87 (1982).